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Carbocyclization

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Rhodium-Catalyzed Regio- and Enantioselective Intermolecular [4+2] Carbocyclization of 4-Alkynals with N,N-Dialkyl Acrylamides**

Ken Tanaka,* Yuji Hagiwara, and Keiichi Noguchi

Transition-metal-catalyzed cycloaddition is a powerful synthetic method for the construction of cyclic frameworks.^[1] In particular, [4+2] cycloadditions of dienes or enynes with alkynes or alkenes have been widely examined for the construction of six-membered-ring systems.^[2-4] However, [4+2] cycloadditions via cyclic acyl metal intermediates to give six-membered carbonyl compounds are rare.^[5] Furthermore, although some transition-metal-catalyzed enantioselective intramolecular [4+2] cycloadditions have been reported,^[3] enantioselective intermolecular [4+2] cycloaddition has not been established.

A recent report described the rhodium-catalyzed [4+2] carbocyclization of 4-alkynals with alkynes to afford cyclo-

[*] Prof. Dr. K. Tanaka, Y. Hagiwara
Department of Applied Chemistry
Graduate School of Engineering
Tokyo University of Agriculture and Technology
Koganei, Tokyo 184-8588 (Japan)
Fax: (+81) 42-388-7037
E-mail: tanaka-k@cc.tuat.ac.jp
Prof. Dr. K. Noguchi
Instrumentation Analysis Center
Tokyo University of Agriculture and Technology
Koganei, Tokyo 184-8588 (Japan)

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- Supporting information for this article (full procedures and characterization data) is available on the WWW under http://www.angewandte.org or from the author.



hexenones via five-membered acyl rhodium intermediates \mathbf{A} , formed through the intramolecular hydroacylation of alkynes. Murakami et al. also reported that similar five-membered acyl rhodium intermediates, formed by insertion of rhodium between the carbonyl carbon atom and the α -carbon atom of cyclobutanones, undergoes intramolecular reaction with alkenes to afford bicyclic cyclohexanones. Herein we reveal that a cationic rhodium(I) complex of (R,R)-walphos $(R,R-7)^{[8]}$ catalyzes a highly regio- and enantioselective intermolecular [4+2] carbocyclization of 4-alkynals with N,N-dialkyl acrylamides to give enantioenriched cyclohexanones (Scheme 1).

$$\begin{array}{c}
O \\
H \\
R
\end{array}$$

$$\begin{array}{c}
Rh^{+} \\
R
\end{array}$$

$$\begin{array}{c}
P \\
R
\end{array}$$

$$\begin{array}{c}
P \\
R
\end{array}$$

Scheme 1. Regio- and enantioselective intermolecular [4+2] carbocyclization of 4-alkynals with alkenes.

We first examined the [4+2] carbocyclization of 5-phenyl-4-pentynal (1a) with alkenes. After screening various alkenes and rhodium(i) complexes, we found that the reaction of 1a with N,N-dimethylacrylamide (2a) in the presence of catalytic [Rh(cod)₂]BF₄/dppb furnished cyclohexanone 3aa in low yield along with alkynyl ketone 4 (Scheme 2). On the other hand, the use of dppe as a ligand did not furnished cyclohexanone 3aa at all.

Scheme 2. [4+2] Carbocyclization of 5-phenyl-4-pentynal (1 a) with N,N-dimethylacrylamide (2 a).

The enantioselective [4+2] carbocyclization of $\mathbf{1a}$ with $\mathbf{2a}$ was examined with various chiral bidentate phosphine ligands (Table 1), which have large P-M-P natural bite angles to facilitate the reductive elimination step. The study revealed that the use of ligand (R,R)-7 dramatically increased the yield and ee value of $\mathbf{3aa}$ (Table 1, entry 5). Although lower yield and ee were observed, the reaction can be carried out with a low catalyst loading of 5% (Table 1, entry 7). The reaction did not proceed at all in the presence of (R,R)-6, which bears a coordinating dimethylamino group (Table 1, entry 4).

A series of 5-substituted-4-pentynals **1a–f** was subjected to the above optimized reaction conditions (Table 2). The reactions of aryl- (Table 2, entries 1–4), alkenyl- (Table 2, entry 6), and alkyl- (Table 2, entry 7) substituted 4-pentynals with **2a** afforded the corresponding cyclohexanones in good yield with high enantioselectivity. The absolute configuration

Table 1: Screening of ligands 5–7 in the rhodium-catalyzed enantioselective [4+2] carbocyclization of 1 a with 2 a to give 3 aa.

Entry	Ligand	Catalyst [%]	Yield [%]	ee [%]
1	(S,S)-diop	20	10 ^[a]	37
2	(R)-Tol-binap	20	9 ^[a]	2
3	(R,S)- 5	20	15 ^[a]	20
4	(R,S)- 6	20	O ^[a]	
5	(R,R)-7	20	72 ^[b]	>99
6	(R,R)-7	10	57 ^[b]	> 99
7	(R,R)-7	5	52 ^[b]	98

[a] Yield determined by NMR spectroscopy. [b] Yields of isolated products.

Table 2: Rhodium-catalyzed enantioselective [4+2] carbocyclization of 5-substituted-4-pentynals with N,N-dialkylacrylamides. [8]

Entry	1 (R)	Alkene	Product	Yield [%] ^[b]	ee [%]
1	a (Ph)	2a	(+)-3 aa	57	> 99
2	b (4-MeOC ₆ H ₄)	2a	(+)-3 ba	58	>99
3	c $(4-F_3CC_6H_4)$	2a	(+)-3 ca	51	94
4	$d (4-CIC_6H_4)$	2a	(+)-(S)-3 da	68	>99
5	$d (4-CIC_6H_4)$	2b	(+)-3 db	77	>99
6 ^[c]	e (1-cyclohexenyl)	2a	(+)-3 ea	50	97
7 ^[c,d]	f (<i>n</i> Bu)	2 a	(+)-(S)- 3 fa	77	>99

[a] No signals for other regioisomers were observed in the ¹H NMR spectra. The *ee* values were determined by chiral HPLC and GC analysis. [b] Yields of isolated products. [c] Catalyst: 20%. [d] Temperature: 40°C.

of (+)-3 da was determined to be S by an anomalous dispersion method (Figure 1),^[9] and that of (+)-3 fa was determined to be S by the circular dichroism exciton chirality method.

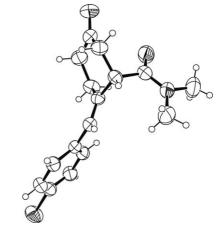


Figure 1. ORTEP diagram of (S)-(+)-3 da. Ellipsoids drawn at the 50% probability level.

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The enantioselective [4+2] carbocyclization of 2-alkynyl benzaldehydes **1g-k** with **2a** was investigated next (Table 3). Aryl- (Table 3, entries 1 and 3), alkenyl- (Table 3, entry 4), and alkyl- (Table 3, entries 5 and 6) substituted 2-ethynyl-

Table 3: Rhodium-catalyzed enantioselective [4+2] carbocyclization of 2-alkynylbenzaldehydes with N,N-dialkylacrylamides. [a]

Entry	1 (R)	Alkene	Product	Yield [%] ^[b]	ee [%]
1	g (Ph)	2a	(+)-3 ga	80	97
2	g (Ph)	2b	(+)-3 gb	80	97
3	h (2-ClC ₆ H ₄)	2a	(+)-3 ha	94	90
4	i (1-cyclohexenyl)	2a	(+)-3 ia	56	99
5	j (nBu)	2a	(+)-3 ja	56	>99
6	k (CI(CH ₂) ₃)	2 a	(+)-3 ka	81	97

[a] No signals for other regioisomers were observed in the ¹H NMR spectra. The *ee* values were determined by chiral HPLC analysis. [b] Yields of isolated products.

benzaldehydes afforded the corresponding cyclohexanones in good yield with high enantioselectivity. Not only *N*,*N*-dimethylacrylamide (**2a**), but also *N*-acryloylpyrrolidine (**2b**) can be employed (Table 2, entry 5 and Table 3, entry 2). In general, the reactions of 2-alkynylbenzaldehydes **1g-k** are more facile than those of 5-substituted-4-pentynals **1a-f**, which allows a lower catalyst loading. Because the [4+2] carbocyclization of 2-alkynyl benzaldehydes with alkynes did not proceed at all, this successful [4+2] carbocyclization of 2-alkynyl benzaldehydes with *N*,*N*-dialkylacrylamides is noteworthy. Importantly, these [4+2] carbocyclizations of 4-alkynals **1a-k** with **2a,b** are highly regioselective and no regioisomers were detected in the crude reaction mixtures. [11]

Scheme 3 shows a plausible mechanism for this carbocyclization. We infer that the rhodium catalyst oxidatively inserts into the aldehyde C–H bond to afford a rhodium acyl hydride **B**. The *cis* addition of the rhodium hydride species to the metal-bound alkyne then provides the five-membered acyl rhodium intermediate \mathbf{A} . Complexation of the alkene is followed by insertion to form metallacycle \mathbf{C} . Reductive elimination furnishes the cyclohexanone $\mathbf{3}$ and regenerates the Rh catalyst. On the other hand, β -hydride elimination from metallacycle \mathbf{C} can furnish the alkynyl ketone $\mathbf{4}$. Alternatively, carbometalation of intermediate \mathbf{B} with the alkene prior to hydrometalation can also furnish $\mathbf{4}$.

In conclusion, we have developed a rhodium-catalyzed regio- and enantioselective intermolecular [4+2] carbocyclization of 4-alkynals with *N*,*N*-dialkylacrylamides. This method serves as an attractive new route to enantioenriched cyclohexanones in view of the one-step access to 4-alkynals from readily available terminal alkynes (1,4-addition to acrolein or Sonogashira coupling with 2-bromobenzaldehyde).

Scheme 3. Plausible mechanism for the rhodium-catalyzed [4+2] carbocyclization of 4-alkynals with N,N-dialkylacrylamides.

Experimental Section

Representative procedure (Table 3, entry 1): Under an Ar atmosphere, a solution of (R,R)-7 (9.9 mg, 0.015 mmol) in CH₂Cl₂ (1.0 mL) was added to a solution of [Rh(cod)₂]BF₄ (6.6 mg, 0.015 mmol) in CH₂Cl₂ (1.0 mL) at room temperature, and the mixture was stirred for 5 min. H₂ was introduced to the resulting solution in a Schlenk tube. The mixture was stirred at room temperature for 0.5 h, and the resulting solution was concentrated to dryness. Benzaldehyde 1g (61.9 mg, 0.300 mmol) and 2a (29.7 mg, 0.300 mmol) were added to the residue by using (CH₂Cl)₂ (1.5 mL). The mixture was stirred at 80°C for 20 h. The resulting solution was concentrated and purified by preparative TLC (hexane/EtOAc 1:4) to furnish (+)-3 ga (73.3 mg, 0.240 mmol, 80% yield, 97% ee) as an orange solid. M.p. 56-58°C; $[\alpha]_{D}^{25}$ +201.0° (acetone, c 1.534; 97% ee); IR (neat): $\tilde{v} = 2930$, 1680, 1630, 1600, 1445, 1400, 1282, 1240, 1142, 772, 758, 730, 700 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) $\delta = 8.12-8.02$ (m, 1H), 7.37–7.28 (m, 1 H), 7.27–7.07 (m, 7 H), 6.67 (s, 1 H), 4.04 (dd, J = 6.9 and 4.5 Hz, 1H), 3.25 (dd, J = 17.4 and 6.9 Hz, 1H), 3.20 (s, 3H), 2.94 (s, 3H), 2.87 ppm (dd, J = 17.4 and 4.5 Hz, 1 H); ¹³C NMR (CDCl₃, 75 MHz) $\delta = 196.2, 170.2, 137.5, 136.1, 134.1, 133.2, 132.2, 129.1, 128.9, 128.7,$ 128.5, 128.3, 127.5, 126.7, 48.7, 43.1, 37.6, 36.0 ppm; HRMS (EI): calcd for $C_{17}H_{13}O$ [M-CONMe₂]⁺: 233.0967; found; 233.0984; HPLC: $t_R =$ 18.3 min (major isomer) and 22.7 min (minor isomer) (CHIRALPAK AD, hexane/iPrOH 90:10, 1.0 mLmin⁻¹)

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- [9] CCDC-277585 (3da) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [10] The use of substituted *N*,*N*-dialkyl acrylamides (*N*,*N*-dimethylmethacrylamide, 1-pyrrolidinylpropenone, *N*-methylmaleimide) did not afford the desired carbocyclization products.
- [11] In the case of the reactions of **1a-k** with Rh¹–(*R,R*)-walphos, no vinyl protons other than those for the desired cyclohexanones **3** were observed by ¹H NMR spectroscopic analysis of the crude reaction mixtures. Although we examined a wide variety of phosphine ligands, the other regioisomer derived from **1a** and **2a** was not obtained at all. On the other hand, the other regioisomer derived from **1g** and **2a** was isolated in trace amounts when

- using Rh¹⁺-dppb. The desired major regioisomer **3ga** and the minor regioisomer can easily be distinguished by the ¹H NMR chemical shifts of their vinyl protons (major isomer: $\delta = 6.68$ ppm (s), minor isomer: $\delta = 6.78$ ppm (d, J = 1.2 Hz)).
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